

Crystal structure of triammonium heptafluorogermanate, $(\text{NH}_4)_3\text{GeF}_7$

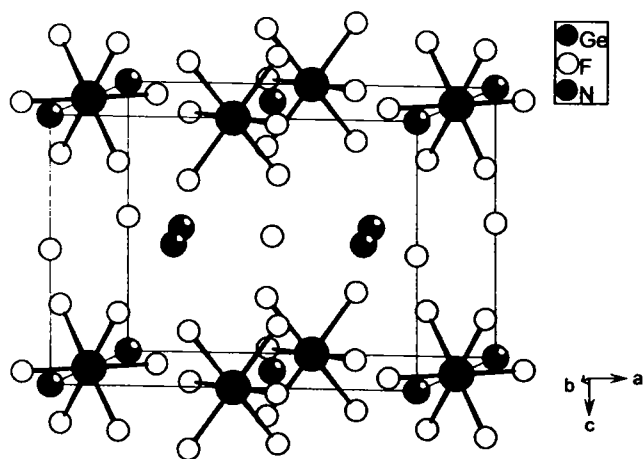
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Received February 4, 1998, transferred to 2nd update of database ICSD in 1998, CSD-No. 404261



$\text{F}_7\text{GeH}_{12}\text{N}_3$, tetragonal, $P4/mbm$ (No. 127), $a = 8.210(1) \text{ \AA}$,
 $c = 5.984(1) \text{ \AA}$, $V = 403.3 \text{ \AA}^3$, $Z = 2$, $R(F) = 0.055$, $R_w(F^2) = 0.140$.

Table 1. Parameters used for the X-ray data collection

Crystal:	colorless, irregular, size 0.1 x 0.1 x 0.1 mm
Wavelength:	Mo $K\alpha$ radiation (0.71073 Å)
μ :	38.68 cm ⁻¹
Diffractometer:	Stoe IPDS
Scan mode:	50 exposures, $\Delta\phi = 3^\circ$
$T_{\text{measurement}}$:	293 K
$2\theta_{\text{max}}$:	54.94°
$N(hkl)_{\text{unique}}$:	275
Criterion for I_0 :	$I_0 > 2 \sigma(I_0)$
$N(\text{param})_{\text{refined}}$:	23
Programs:	SHELXS-86, SHELXL-93

Source of material: Colorless single crystals of $(\text{NH}_4)_3\text{GeF}_7$ were obtained during the reaction of Ge powder with $(\text{NH}_4)\text{HF}_2$ (molar ratio 1:4) in a sealed Monel ($\text{Cu}_{32}\text{Ni}_{68}$) ampoule at 573 K. At higher temperature ($> 673 \text{ K}$), $(\text{NH}_4)\text{GeF}_6$ (see ref. 2) is the main product.

The crystal structure of triammonium heptafluorogermanate contains as the isotopic $(\text{NH}_4)_3\text{SiF}_7$ (see ref. 3) isolated $[\text{GeF}_6]$ octahedra [$d(\text{Ge}^{4+}-\text{F}^-) = 175.6 \text{ pm}$ (F2); 175.8 pm (F3)] and lonesome F⁻ ions (F1).

Table 2. Final atomic coordinates and displacement parameters (in Å²)

Atom	Site	x	y	z	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
Ge	2d	0	1/2	0	0.0433(7)	U_{11}	0.0373(8)	-0.0031(6)	0	0
F(1)	2b	0	0	1/2	0.075(4)	U_{11}	0.079(7)	0	0	0
F(2)	4g	0.3488(8)	$x+1/2$	0	0.195(9)	U_{11}	0.113(8)	-0.13(1)	0	0
F(3)	8k	0.1008(6)	$x+1/2$	0.219(1)	0.151(5)	U_{11}	0.099(4)	-0.054(6)	-0.041(3)	U_{13}
N(1)	2a	0	0	0	0.10(1)	U_{11}	0.18(3)	0	0	0
N(2)	4h	0.293(1)	$x+1/2$	1/2	0.104(5)	U_{11}	0.050(5)	-0.044(7)	0	0

Acknowledgment. We thank the Deutsche Forschungsgemeinschaft for financial support.

References

- Plitzko, C.: Neue komplexe Ammoniumfluoride und Fluorid-Ammoniakate durch Umsetzung von Metallen mit Ammonium- und Hydraziniumfluoriden. Dissertation, Universität Hannover, Germany 1996.
- Vajnstejn, B. K.; Kurdjumova, R. N.: Cubic Modification of $(\text{NH}_4)_2\text{GeF}_6$. *Sov. Phys. Cryst.* **3** (1958) 27-30.
- Hoard, L.; Williams, M. B.: Structures of Complex Fluorides. Ammonium Hexafluorosilicate - Ammonium Fluoride, $(\text{NH}_4)_2\text{SiF}_6 \cdot \text{NH}_4\text{F}$. *J. Am. Chem. Soc.* **64** (1942) 633-637.
- Sheldrick, G. M.: Phase Annealing in SHELX-90: Direct Methods for Large Structures. *Acta Crystallogr.* **A46** (1990) 467-473.
- Sheldrick, G. M.: SHELXL-93. Program for refining crystal structures. University of Göttingen, Germany 1993.