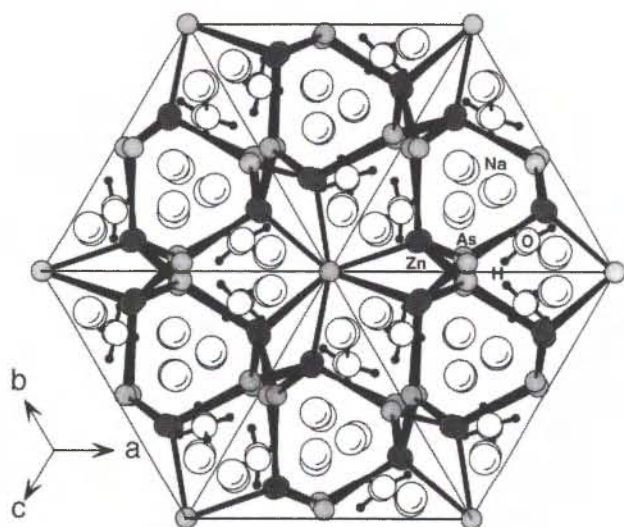


Crystal structure of hexasodium trizinc tetraarsenate trihydrate, $\text{Na}_6\text{Zn}_3(\text{AsO}_4)_4 \cdot 3\text{H}_2\text{O}$

T. M. Gesing*^I and R. Wartchow^{II}^I Universität Hannover, Institut für Mineralogie, Welfengarten 1, D-30167 Hannover, Germany^{II} Universität Hannover, Institut für Anorganische Chemie, Callinstr. 9, D-30167 Hannover, Germany

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Abstract

$\text{As}_4\text{H}_6\text{Na}_6\text{O}_{19}\text{Zn}_3$, cubic, $P2_13$ (No. 198), $a = 12.245(1) \text{ \AA}$,
 $V = 1836.0 \text{ \AA}^3$, $Z = 4$, $R_{\text{gt}}(F) = 0.018$, $R_w(F^2) = 0.036$, $T = 300 \text{ K}$.

Source of material

The material was crystallised from NaOH, ZnO, H_3AsO_4 and H_2O under hydrothermal conditions at 533 K with autogeneous pressure for 130 h in a teflon coated steel autoclave under static conditions.

Table 3. Atomic coordinates and displacement parameters (in Å^2).

Atom	Site	x	y	z	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
As(1)	4a	0.03146(2)	x	x	0.0055(1)	U_{11}	U_{11}	0.0000(1)	U_{12}	U_{12}
As(2)	4a	0.26905(3)	x	x	0.0047(1)	U_{11}	U_{11}	0.0005(1)	U_{12}	U_{12}
As(3)	4a	0.51778(2)	x	x	0.0051(1)	U_{11}	U_{11}	0.0004(1)	U_{12}	U_{12}
As(4)	4a	0.77878(3)	x	x	0.0056(1)	U_{11}	U_{11}	0.0002(1)	U_{12}	U_{12}
Zn(1)	12b	0.12491(3)	0.49148(3)	0.24714(3)	0.0061(2)	0.0073(2)	0.0073(2)	0.0004(1)	0.0014(1)	0.0002(1)
Na(1)	12b	0.9943(1)	0.2431(1)	0.1722(1)	0.0188(7)	0.0110(6)	0.0183(7)	-0.0010(5)	0.0019(6)	-0.0011(5)
Na(2)	12b	0.2457(1)	0.4206(1)	0.5020(1)	0.0196(8)	0.0225(7)	0.0099(6)	-0.0026(5)	-0.0028(5)	-0.0006(5)

Discussion

The compound was described by Grey et al. [1] together with its relation to garnet. The crystal structure is built up by ZnO_4 and AsO_4 tetrahedra. In the figure, the oxygen atoms are not drawn for clarity. The sodium atoms are octahedrally coordinated by five tetrahedra oxygen atoms and the oxygen atoms of the water molecule (O9). The hydrogen atoms form hydrogen bridging bonds to the two tetrahedra oxygen atoms O2 (H91: 191.8 pm) and O4 (H92: 199.3 pm). The mean metal-oxygen distances are 248 pm for Na1 and Na2, 194 pm for Zn1 and 168 pm – 169 pm for the different As atoms.

Table 1. Data collection and handling.

Crystal:	colorless, cube-shaped, size 0.10 x 0.10 x 0.10 mm
Wavelength:	Mo $K\alpha$ radiation (0.71073 Å)
μ :	132.96 cm^{-1}
Diffractometer, scan mode:	Stoe IPDS, 140 exposures, $\Delta\phi = 1.7^\circ$
$2\theta_{\text{max}}$:	56.5°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	20984, 1508
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 1338
$N(\text{param})_{\text{refined}}$:	105
Programs:	STRUCTURE TIDY [2], SHELXS-86 [3], SHELXL-93 [4], DIAMOND [5]

Table 2. Atomic coordinates and displacement parameters (in Å^2).

Atom	Site	x	y	z	U_{iso}
H(91)	12b	0.035(4)	0.179(2)	0.375(5)	0.06(1)
H(92)	12b	0.068(3)	0.299(4)	0.389(5)	0.06(1)

* Correspondence author

(e-mail: tm.gesing@mineralogie.uni-hannover.de)

Table 3. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
O(1)	12 <i>b</i>	0.0099(2)	0.5440(2)	0.3380(2)	0.012(1)	0.021(1)	0.011(1)	0.006(1)	0.006(1)	0.003(1)
O(2)	12 <i>b</i>	0.0550(2)	0.4205(2)	0.1187(2)	0.013(1)	0.009(1)	0.009(1)	0.0013(9)	-0.0039(9)	0.0001(9)
O(3)	12 <i>b</i>	0.1940(2)	0.2292(2)	0.5953(2)	0.015(1)	0.017(1)	0.013(1)	0.006(1)	-0.0038(9)	-0.008(1)
O(4)	12 <i>b</i>	0.1981(2)	0.3735(2)	0.3263(2)	0.014(1)	0.010(1)	0.009(1)	0.0052(9)	0.0022(9)	-0.0017(9)
O(5)	4 <i>a</i>	0.1916(2)	<i>x</i>	<i>x</i>	0.0134(8)	<i>U</i> ₁₁	<i>U</i> ₁₁	-0.0020(9)	<i>U</i> ₁₂	<i>U</i> ₁₂
O(6)	4 <i>a</i>	0.4397(2)	<i>x</i>	<i>x</i>	0.0149(9)	<i>U</i> ₁₁	<i>U</i> ₁₁	-0.0026(9)	<i>U</i> ₁₂	<i>U</i> ₁₂
O(7)	4 <i>a</i>	0.7009(2)	<i>x</i>	<i>x</i>	0.0148(8)	<i>U</i> ₁₁	<i>U</i> ₁₁	-0.0031(9)	<i>U</i> ₁₂	<i>U</i> ₁₂
O(8)	4 <i>a</i>	0.9535(2)	<i>x</i>	<i>x</i>	0.0150(8)	<i>U</i> ₁₁	<i>U</i> ₁₁	-0.0037(9)	<i>U</i> ₁₂	<i>U</i> ₁₂
O(9)	12 <i>b</i>	0.0046(2)	0.2523(2)	0.3764(2)	0.024(2)	0.022(2)	0.026(1)	-0.001(1)	-0.000(1)	0.002(1)

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