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Crystal structure of (1*S*,2*R*,5*R*,6*R*)-2-methoxy-6-(16-nitrophenylethynyl)-1-azabicyclo[3.2.2]nonane, C₁₇H₂₀N₂O₃

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Abstract

 $C_{17}H_{20}N_2O_3$, orthorhombic, $P_{21}2_12_1$ (No. 19), a = 7.096(1) Å, b = 11.671(1) Å, c = 19.277(2) Å, V = 1596.5 Å³, Z = 4, $R_{gt}(F) = 0.032$, $wR_{ref}(F^2) = 0.040$, T = 300 K.

Source of material

p-NO₂-Phenyl substituted 10,11-didehydroquincorine [1] was transformed into its corresponding C9-iodide via mesylation and Lil-mediated iodination. The resulting 2-iodomethyl-5-(16-nitrophenylethynyl)-1-azabicyclo[3.2.2]-nonane (500 mg, 1.26 mmol, 1 eq) was dissolved in abs. MeOH (6 ml). After being stirred for 5 min at room temperature under argon freshly prepared silver benzoate (318 mg, 1.39 mmol, 1.1 eq) was added. The reaction mixture was stirred for 16 h at 323 K, diluted with MeOH and filtered. The remaining precipitate was washed with MeOH, the combined solutions were concentrated in vacuo and the resulting solid was dissolved in CH₂Cl₂. After addition of saturated aqueous NaHCO3, the aqueous layer was extracted with CH₂Cl₂, and the combined organic layer dried (MgSO₄) and evaporated. Purification by chromatography (MTBE) furnished the desired substituted 1-azabicyclo[3.2.2]nonane (77 %, 292 mg, 0.97 mmol) which was recrystallized from MTBE / PE.

Discussion

Azabicyclic systems are common structural units in medicinal chemistry and in a large number of natural products, especially alkaloids including cinchona species [2]. Although azabicyclo[2.2.1]heptane and azabicyclo[2.2.2]octane systems have been subject to intensive studies, the next higher [3.2.2]-homologues have hardly been investigated [3-5]. Due to uncertain stereochemical and mechanistic details of the hetero-cinchona rearrangement these reactions were reinvestigated [6]. The resulting quinine and quinidine derivatives contain a 1-azabicyclo[3.2.2]nonane moiety. The mechanism of the rearrangement involves a 1,2-nucleophilic shift with generation of a strained non-planar iminium-ion and a stereoselective nucleophilic capture by the solvent. Likewise, silver salt-mediated ring enlargement of halogenated Quincorine® and Quincoridine® derivatives offers a short and stereoselective route to novel substituted 1-azabicyclo[3.2.2]nonanes containing four stereogenic centres [7]. In view of the high synthetic and therapeutic potential of 5-substituted quinuclidine derivatives we prepared (1S,2R,5R,6R)-2-methoxy-6-(16-nitrophenylethynyl)-1-azabicyclo[3.2.2]nonane and present its crystal structure. Small torsion angles Φ_1 (N1-C7-C6-C5) = 6.3° and Φ_2 $(N1-C8-C9-C5) = 2.4^{\circ}$ suggest reduced conformational strain of

the 1-azabicyclo[3.2.2]nonane moiety compared with related 1-azabicyclo[2.2.2]octanes which are strongly twisted to minimize eclipsing and ring strain [1]. The bridgehead nitrogen in the title compound is more flattened, with an expansion of bond angles, than in quincorine derivatives [8].

Table 1. Data collection and handling.

Crystal:	colourless, plate (010), size 0.09 × 0.18 × 0.33 mm
Wavelength: µ: Diffractometer, scan mode: $2\theta_{max}$: $N(hkl)_{measured}$, $N(hkl)_{unique}$: Criterion for I_{obs} , $N(hkl)_{gl}$: $N(param)_{refined}$: Programs:	Mo K_{α} radiation (0.71073 Å) 0.86 cm ⁻¹ Stoe IPDS, 158 exposures, $\Delta \phi = 1.3^{\circ}$ 48.24° 10621, 2526 $I_{obs} > 2 \sigma(I_{obs})$, 1139 199 SHELXS-86 [9], SHELXL-93 [10], PLATON [11]

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Atom	Site	x	у	z	$U_{\rm iso}$
		0.0(45(4)	0.0001(0)	0.000((0)	0.070
H(1)	4a	0.8645(4)	0.2091(2)	0.2996(2)	0.078
H(2)	4a	0.9639(4)	0.3901(2)	0.3272(2)	0.095
H(3)	4a	0.8505(4)	0.3900(2)	0.3969(2)	0.095
H(4)	4a	0.6946(4)	0.4082(2)	0.2630(2)	0.095
H(5)	4 <i>a</i>	0.6902(4)	0.5036(2)	0.3199(2)	0.095
H(6)	4a	0.3967(5)	0.4448(2)	0.3258(2)	0.091
H(7)	4a	0.2800(4)	0.2637(2)	0.3115(2)	0.087
H(8)	4a	0.4396(4)	0.1080(2)	0.3298(2)	0.086
H(9)	4 <i>a</i>	0.5918(4)	0.1412(2)	0.2745(2)	0.086
H(10)	4 <i>a</i>	0.6704(4)	0.2547(2)	0.4655(2)	0.097

Table 2. Atomic coordinates and displacement parameters (in $Å^2$).

Table 2. Continued.

Atom	Site	<i>x</i>	У	Z	Uiso
H(11)	4 <i>a</i>	0.4831(4)	0.1880(2)	0.4506(2)	0.097
H(12)	4a	0.3430(5)	0.3509(2)	0.4272(2)	0.121
H(13)	4a	0.5329(5)	0.4171(2)	0.4387(2)	0.121
H(14)	4a	1.1278(4)	0.0556(2)	0.4115(2)	0.161
H(15)	4a	1.0614(4)	0.0639(2)	0.3341(2)	0.161
H(16)	4a	0.9157(4)	0.0347(2)	0.3926(2)	0.161
H(17)	4a	0.3980(4)	0.4780(2)	0.0721(2)	0.075
H(18)	4a	0.3938(4)	0.4789(2)	-0.0471(2)	0.077
H(19)	4a	0.4085(4)	0.1353(3)	-0.0475(2)	0.086
H(20)	4a	0.4102(4)	0.1350(3)	0.0709(2)	0.093

Table 3. Atomic coordinates and displacement parameters (in $Å^2$).

Atom	Site	<i>x</i>	у	<i>z</i>	<i>U</i> ₁₁	U ₂₂	U33	U ₁₂	<i>U</i> ₁₃	U ₂₃
O (1)	4 <i>a</i>	0.9889(3)	0.1968(2)	0.3905(1)	0.069(1)	0.087(1)	0.117(2)	-0.006(1)	-0.022(1)	0.036(1)
O(2)	4a	0.3836(4)	0.3960(2)	-0.1644(1)	0.213(3)	0.098(2)	0.077(2)	-0.010(2)	-0.022(2)	0.009(2)
O(3)	4a	0.4110(4)	0.2128(2)	-0.1631(1)	0.151(2)	0.103(2)	0.093(2)	0.019(2)	-0.015(2)	-0.032(2)
N(1)	4a	0.6635(3)	0.1844(2)	0.3705(1)	0.066(2)	0.050(1)	0.047(2)	0.003(1)	0.002(1)	0.002(1)
N(2)	4a	0.3999(3)	0.3048(3)	-0.1335(2)	0.092(2)	0.081(2)	0.083(3)	0.005(2)	-0.014(2)	-0.006(2)
C(2)	4a	0.8396(4)	0.2348(2)	0.3471(2)	0.068(2)	0.058(2)	0.068(2)	-0.003(2)	0.001(2)	0.013(2)
C(3)	4 <i>a</i>	0.8479(4)	0.3650(2)	0.3490(2)	0.102(3)	0.052(2)	0.084(3)	-0.016(2)	-0.023(2)	0.009(2)
C(4)	4a	0.6819(4)	0.4216(2)	0.3124(2)	0.114(3)	0.043(2)	0.080(2)	0.005(2)	0.002(3)	0.005(2)
C(5)	4a	0.4841(5)	0.3815(2)	0.3350(2)	0.093(3)	0.064(2)	0.072(2)	0.029(2)	0.010(2)	0.007(2)
C(6)	4 <i>a</i>	0.4099(4)	0.2767(2)	0.2961(2)	0.065(2)	0.075(2)	0.077(2)	0.011(2)	0.005(2)	0.013(2)
C(7)	4 <i>a</i>	0.5264(4)	0.1679(2)	0.3157(2)	0.076(2)	0.064(2)	0.074(2)	0.001(2)	0.000(2)	0.010(2)
C(8)	4 <i>a</i>	0.5746(4)	0.2402(2)	0.4306(2)	0.108(2)	0.073(2)	0.062(2)	0.017(2)	0.017(2)	0.012(2)
C(9)	4 <i>a</i>	0.4739(5)	0.3553(2)	0.4130(2)	0.155(3)	0.082(2)	0.064(2)	0.048(2)	0.027(2)	0.008(2)
C(10)	4a	1.0265(4)	0.0781(2)	0.3814(2)	0.093(3)	0.088(2)	0.221(5)	0.010(2)	0.012(3)	0.067(3)
C(11)	4 <i>a</i>	0.4052(4)	0.2935(3)	0.2202(2)	0.062(2)	0.084(2)	0.083(3)	-0.001(2)	-0.008(2)	0.016(2)
C(12)	4a	0.4047(4)	0.3038(3)	0.1596(2)	0.069(2)	0.077(2)	0.080(3)	0.001(2)	-0.010(2)	0.014(2)
C(13)	4a	0.4047(4)	0.3064(3)	0.0851(2)	0.053(2)	0.061(2)	0.071(2)	-0.002(2)	-0.009(2)	0.002(2)
C(14)	4 <i>a</i>	0.3998(4)	0.4089(2)	0.0481(2)	0.065(2)	0.048(2)	0.075(2)	-0.002(2)	-0.012(2)	-0.009(2)
C(15)	4a	0.3977(4)	0.4101(2)	-0.0228(2)	0.071(2)	0.049(2)	0.072(2)	-0.003(2)	-0.010(2)	-0.001(2)
C(16)	4a	0.4015(3)	0.3065(3)	-0.0576(2)	0.057(2)	0.060(2)	0.057(2)	0.003(2)	-0.009(2)	-0.002(2)
C(17)	4a	0.4062(4)	0.2041(3)	-0.0232(2)	0.083(2)	0.046(2)	0.086(3)	0.007(2)	-0.010(2)	-0.005(2)
C(18)	4a	0.4074(4)	0.2045(3)	0.0473(2)	0.092(2)	0.050(2)	0.089(2)	0.005(2)	-0.009(2)	0.016(2)

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References

- Schrake, O.; Braje, W.; Hoffmann, H. M. R.; Wartchow, R.: Synthesis of 10,11-Didehydro- and 10,11-Dihydro-quincorine and of the Quincoridine analogs: Functionalized and Enantiopure 1-Azabicyclo[2.2.2]octanes with Four Stereogenic centres. Tetrahedron: Asymmetry 9 (1998) 3717-3722.
- Chen, Z.; Trudell, M. L.: Chemistry of 7-Azabicyclo[2.2.1]hepta-2,5-dienes, 7-Azabicyclo[2.2.1]hept-2-enes, and 7-Azabicyclo[2.2.1]heptanes. Chem. Rev. 96 (1996) 1179-1193.
- Davies, H. M. L.; Hodges, L. M.; Thornley, C. T.: Rhodium(II)-catalyzed Decomposition of Vinyldiazomethanes in the Presence of 1,2-Dihydropyridines: Synthesis of the 6-Azabicyclo[3.2.2]nonane Nucleus. Tetrahedron Lett. 39 (1998) 2707-2710.
- Prelog, V.; Cerkovnikov, E.: Über die Synthese des 1-Azabicyclo[3.2.2]nonans, der Chinuclidin-carbonsäure-2 und der β-(γ-Piperidyl)-propionsäure. Liebigs Ann. 83 (1937) 532.

- Pearson, W. H.; Walavalkar, R.; Schkeryantz, J. M.; Fang, W.-K.; Blickensdorf, J. D.: Intramolecular Schmidt Reactions of Azides with Carbocations: Synthesis of Bridged-Bicyclic and Fused-Bicyclic Tertiary Amines. J. Am. Chem. Soc. 115 (1993) 10183-10194.
- Braje, W. M.; Wartchow, R.; Hoffmann, H. M. R.: Struktur und Mechanismus in der Cinchonaalkaloidchemie: Ein 50 Jahre altes Fehlkonzept wird widerlegt. Angew. Chem. 111 (1999) 2698-2701.
- Röper, S.; Frackenpohl, J.; Schrake, O.; Wartchow, R; Hoffmann, H. M. R.: Synthesis of Enantiopure 1-Azavicyclo[3.2.2]nonanes via Stereoselective Capture of Chiral Carbocations. Org. Lett 2 (2000), in press.
- Wartchow, R.; Schrake, O.; Braje, W. M.; Hoffmann, H. M. R.: Crystal structure of (15,25,45,5R)-2-hydroxymethyl-5-ethynyl-1-azabicyclo[2.2.2]octane. Z. Kristallogr. NCS 214 (1999) 285-286.
- Sheldrick, G. M.: SHELXS-86, program for crystal structure determination University of Göttingen, Germany 1986.
- Sheldrick, G. M.: SHELXL-93, program for refining crystal structures University of Göttingen, Germany 1993.
- Spek, A. L.: PLATON, an integrated tool for the analysis of the results of a single crystal structure determination. Acta Crystallogr. Suppl. A46 (1990) C-34.