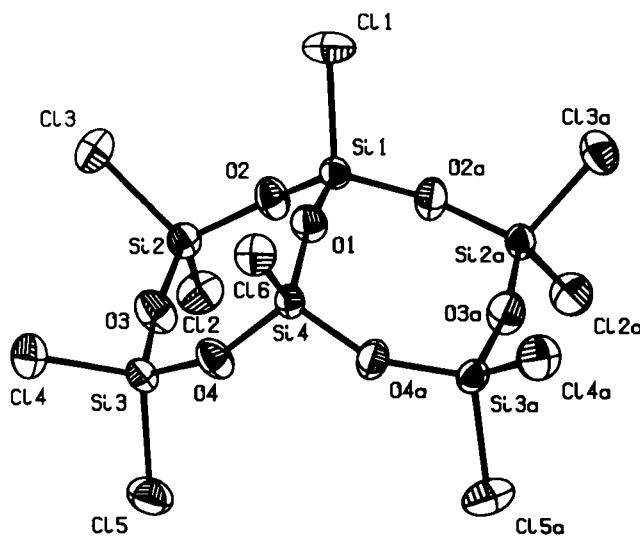


# Crystal structure of decachloro-bicyclo[5.5.1]heptaoxatridesilane, $\text{Si}_6\text{O}_7\text{Cl}_{10}$

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**Abstract**

$\text{Cl}_{10}\text{O}_7\text{Si}_6$ , monoclinic,  $P12_1/m1$  (No. 11),  $a = 6.094(1)$  Å,  $b = 20.676(4)$  Å,  $c = 8.295(2)$  Å,  $\beta = 99.47(3)^\circ$ ,  $V = 1030.9$  Å<sup>3</sup>,  $Z = 2$ ,  $R_{\text{gt}}(F) = 0.039$ ,  $wR_{\text{obs}}(F^2) = 0.114$ ,  $T = 203$  K.

**Source of material**

The general preparation method is described in [1]. Solid  $\text{Si}_6\text{O}_7\text{Cl}_{10}$  could be isolated in small amounts by sublimation of one distillate fraction.

**Table 2.** Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

Atom	Site	x	y	z	$U_{11}$	$U_{22}$	$U_{33}$	$U_{12}$	$U_{13}$	$U_{23}$
Si(1)	2e	-0.2219(2)	1/4	0.3737(1)	0.0331(6)	0.0221(5)	0.0215(6)	0	0.0054(5)	0
Si(2)	4f	-0.0008(1)	0.38307(3)	0.4467(1)	0.0327(5)	0.0249(5)	0.0274(5)	-0.0032(2)	0.0090(4)	0.0013(2)
Si(3)	4f	-0.1247(1)	0.38101(3)	0.80369(9)	0.0332(5)	0.0251(5)	0.0252(5)	-0.0040(2)	0.0070(3)	-0.0043(2)
Si(4)	2e	-0.3613(2)	1/4	0.7220(1)	0.0313(6)	0.0214(5)	0.0243(6)	0	0.0086(5)	0
O(1)	2e	-0.3363(5)	1/4	0.5335(3)	0.044(2)	0.034(1)	0.026(2)	0	0.013(1)	0
O(2)	4f	-0.0678(4)	0.31279(8)	0.3742(3)	0.047(1)	0.026(1)	0.039(1)	-0.0074(8)	0.015(1)	-0.0026(8)
O(3)	4f	-0.0501(4)	0.38869(9)	0.6292(3)	0.057(2)	0.042(1)	0.033(1)	-0.0120(8)	0.020(1)	-0.0043(8)
O(4)	4f	-0.2468(4)	0.31289(8)	0.8115(3)	0.054(2)	0.029(1)	0.032(1)	-0.0132(8)	0.012(1)	-0.0044(7)
Cl(1)	2e	-0.4572(2)	1/4	0.1743(2)	0.0558(8)	0.0728(8)	0.0298(7)	0	-0.0095(5)	0
Cl(2)	4f	0.3241(1)	0.39722(4)	0.4443(1)	0.0308(5)	0.0501(5)	0.0506(6)	-0.0030(3)	0.0077(4)	0.0055(3)
Cl(3)	4f	-0.1772(1)	0.45067(3)	0.3077(1)	0.0403(5)	0.0371(5)	0.0521(6)	0.0057(3)	0.0073(4)	0.0110(3)
Cl(4)	4f	-0.3320(1)	0.45268(3)	0.8382(1)	0.0510(6)	0.0401(5)	0.0535(6)	0.0124(3)	0.0073(4)	-0.0071(3)
Cl(5)	4f	0.1435(1)	0.38397(4)	0.9801(1)	0.0401(5)	0.0621(6)	0.0456(6)	0.0024(3)	-0.0051(4)	-0.0161(3)
Cl(6)	2e	-0.6872(2)	1/4	0.7383(1)	0.0319(6)	0.0467(6)	0.0425(7)	0	0.0122(5)	0

**Discussion**

$\text{Si}_6\text{O}_7\text{Cl}_{10}$  is a structural link between the eight-membered ring compound  $\text{Si}_4\text{O}_4\text{Cl}_8$  [2] and the cube-like silsesquioxane  $\text{Si}_8\text{O}_{12}\text{Cl}_8$  [3].  $\text{Si}_6\text{O}_7\text{Cl}_{10}$  marks one step in the reaction sequence between  $\text{SiCl}_4$  and  $\text{O}_2$  leading to amorphous  $\text{SiO}_2$ . This isomer of  $\text{Si}_6\text{O}_7\text{Cl}_{10}$  is a bicyclic compound consisting of two edge connected eight-membered rings.

**Table 1.** Data collection and handling.

Crystal:	colourless, irregular shape, size $0.2 \times 0.2 \times 0.2$ mm
Wavelength:	Mo $K_\alpha$ radiation ( $0.71073$ Å)
$\mu$ :	$17.20$ cm <sup>-1</sup>
Diffractometer, scan mode:	Stoe IPDS, 100 exposures, $\Delta\varphi = 0.8^\circ$
$2\theta_{\text{max}}$ :	$48^\circ$
$N(hkl)$ measured, $N(hkl)$ unique:	3659, 1341
Criterion for $I_{\text{obs}}$ , $N(hkl)$ gt:	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$ , 1289
$N(\text{param})$ refined:	112
Programs:	SHELXS-86 [4], SHELXL-93 [5], PLATON [6]

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