

Institute of Electrical Engineering and Measurement Technology Department of Sensors and Measurement Technology



Extending the Golay Equation for Coupling a Gas Chromatograph to a Drift Tube IMS

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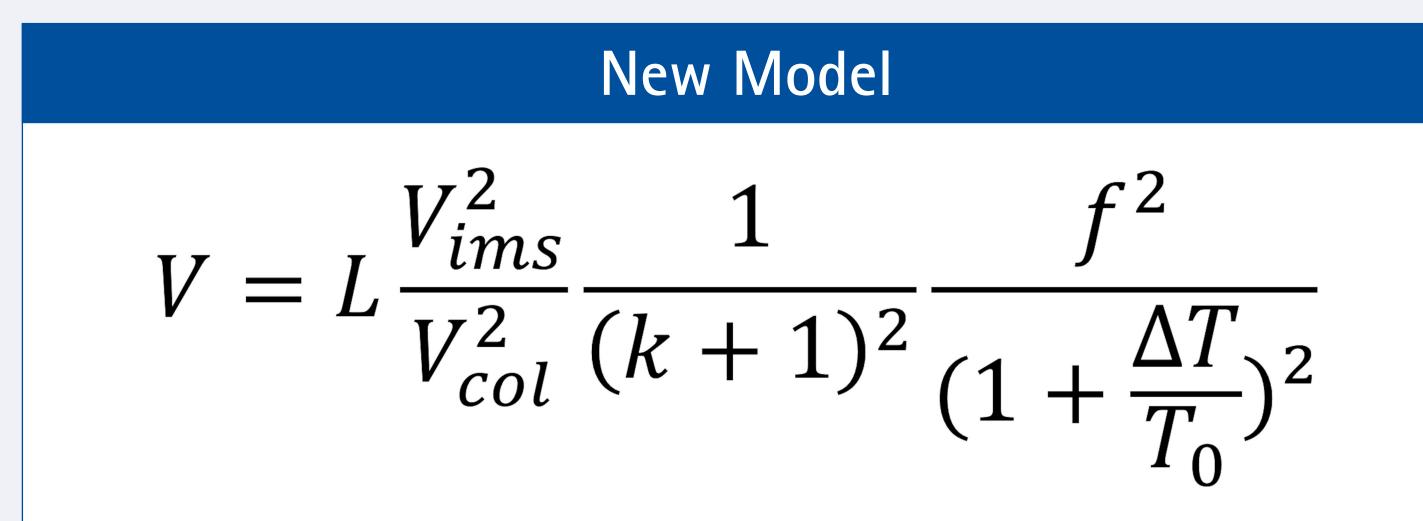
Introduction

Due to chemical cross sensitivities and limits in resolving power of ion mobility spectrometers (IMS), a preseparation of complex mixtures is required. In most cases, gas chromatographic (GC) pre-separation is used [1-3]. In contrast to typical detectors, an IMS adds a significant dead volume to the system which leads to peak

broadening. In the traditional Golay equation [4], the plate heights H are described by the longitudinal diffusion B and the radial diffusion C inside the column. Therefore, we extended this equation by adding a new term V to incorporate the influence of the IMS volume.

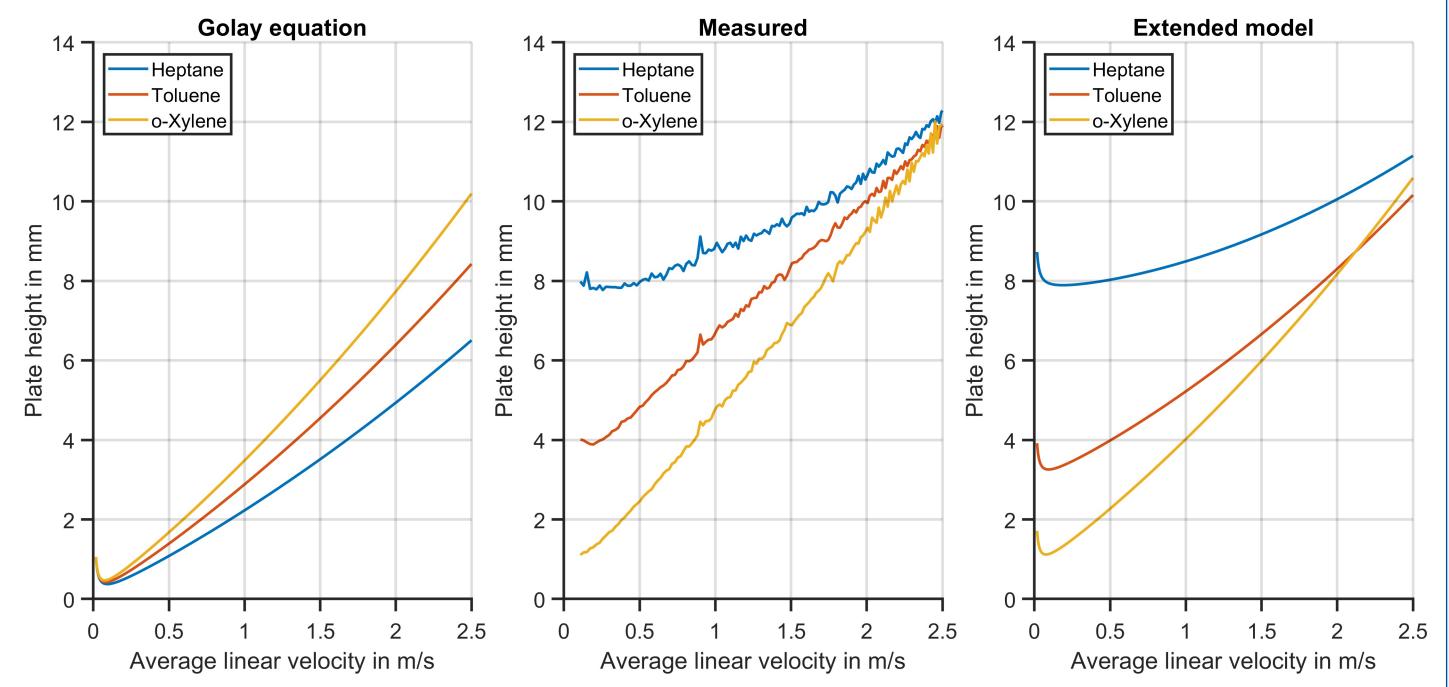
 $H = V + - + C\bar{u}$

Our model has been experimentally verified using an ultra-high sensitive IMS detector in combination with capillary columns of different lengths and diameters.



- » Adds a new term V to the plate height H
- » New term can be divided into three parts:
- » 1: Relation between IMS volume V_{ims} and column volume V_{col}
- » 2: The retention coefficient k of the observed substance





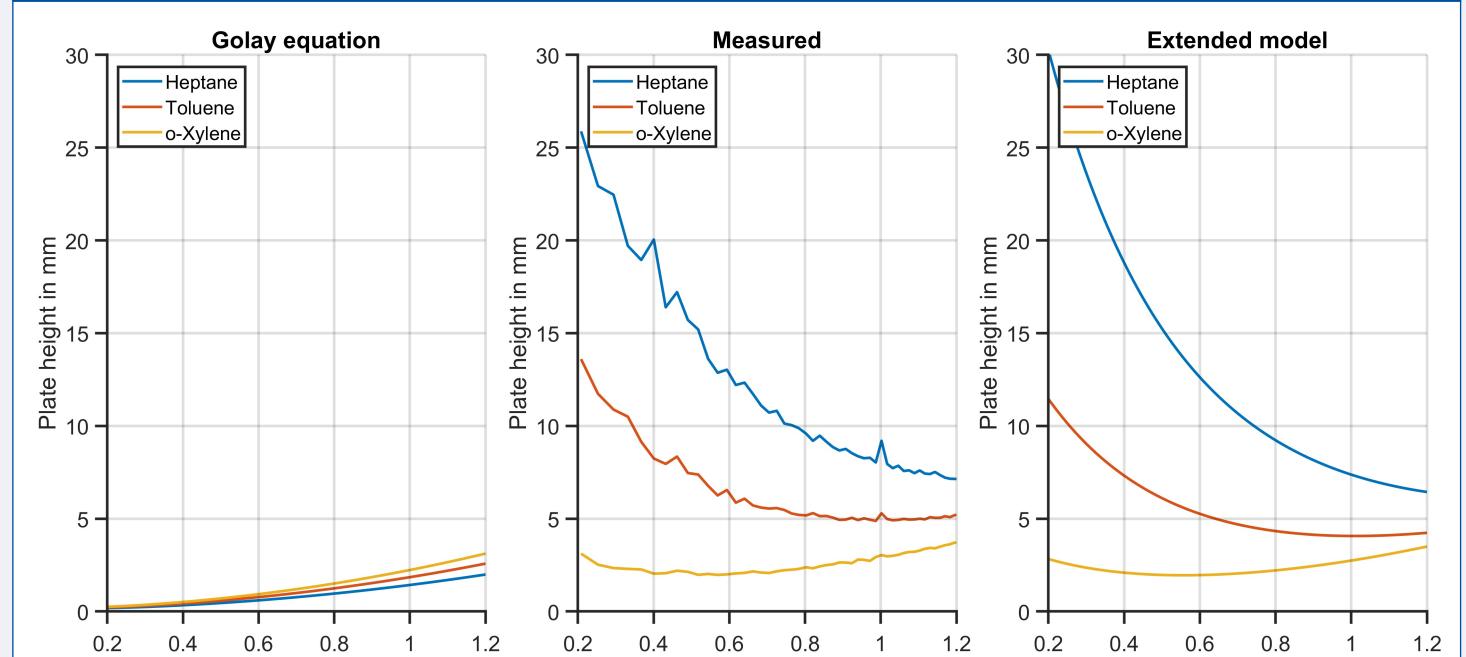
» 3: Gas expansion due to thermal expansion caused by the temperature difference ΔT between IMS and GC and the pressure correction f [5]

$L = 30 \text{ m} C_d = 530 \mu \text{m}$ Golay equation **Extended model** Measured 14 --Heptane Heptane Heptane Toluene Toluene Toluene -o-Xylene o-Xylene o-Xylene 10 in mm mm height in .⊑ height height Plate Plate ate 2.5 0.5 1.5 0.5 1.5 2.5 1.5 2.5 0 Average linear velocity in m/s Average linear velocity in m/s Average linear velocity in m/s

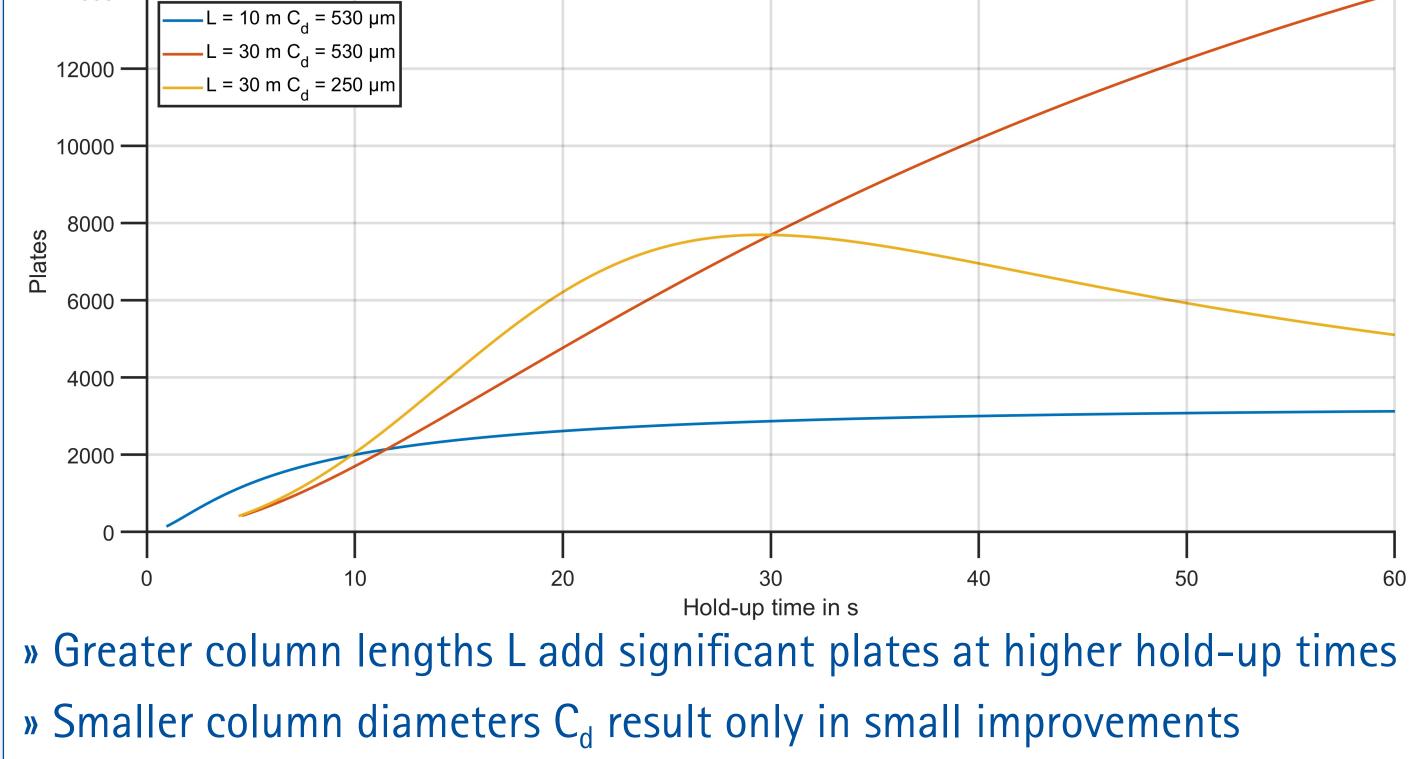
Direct Comparison of the Three Columns

14000			
14000			

- » Left: expected plate heights from the pressure corrected [5] Golay equation
- » Middle: measurement with an actual GC-column at isotherm 70°C and the IMS as detector
- » Right: expected plate heights from the extended Golay equation including the new term V



$L = 30 \text{ m} C_d = 250 \mu \text{m}$



Average linear velocity in m/s Average linear velocity in m/s Average linear velocity in m/s » For a small column volume, plate heights are dominated by the influence of the new term V

Acknowledgments & References

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[1] Kanu, Abu B., J. Chrom. A (1177), pp. 12-27 [2] Allers, M., Journal of Breath Research 10 (2016) [3] Garrido-Delgado, Food Chemistry 187 (2015), pp. 572–579 [4] M. J. E. Golay, Gas Chromatography (1958), pp. 36–55 [5] Giddings, J. C., Analytical chemistry 32 (1960) 7, pp. 867–870