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[Plen2]

Are MOF membranes better than those made of zeolites?

J. Caro*

Leibniz University Hannover, Germany

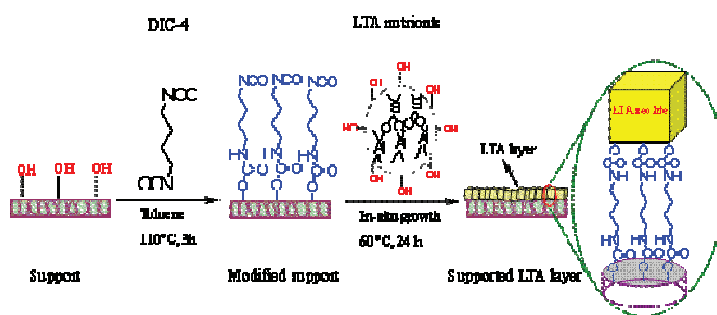
During the last few years, remarkable progress in the development of zeolite and MOF membranes have been obtained. Examples in the field of zeolite membranes are the development of LTA molecular sieve membranes as cation-containing aluminumsilicate [1] and cation-free aluminiumphosphate [2] and ITQ-29 silica [3]. Examples in the field of MOF (metal organic framework) membranes are the ZIF (zeolitic imidazolate framework) membranes ZIF-7 [4], 8 [5], 22 [6] and 90 [7], all of them with hydrogen selectivity.

The above addressed progress in the development of zeolite and MOF membranes could be achieved by the rigorous application of sophisticated new and improved synthesis tools:

- Matching of the zeta potential between support and the precursor particles in solution: By docking of modifying molecules or control of the pH, a surface charge which electrostatically attracts seeds or precursors in a seeding-free synthesis, can be established.
- Application of microwave heating in zeolite and MOF membrane synthesis: In combination with seeding, the quick heating up of the synthesis solution suppresses new nucleation, only the attached to the support seeds grow.
- Development of new seeding techniques using zeolite and MOF nano seeds: By using PEI (polyetherimine), seeds become fixed via H-bridging to the support.
- Covalent bonding between the growing zeolite and MOF layer and the support: By using di-socyanate or APTES (aminopropyltriethoxysilane), first the surface charge can be tuned, and second a covalent bond between the support and the seed/molecular sieve layer can be formed.

Covalent seeds to by using linker

(A. Mater. 1142)



bonding of LTA an alumina support di-isocyanate as

Huang , J. Caro, J. Chem. 21 (2011)

Whereas

for zeolite and MOF membranes are very similar, there exist severe differences between them:

- Because of the phenomenon framework flexibility, MOF membranes show no sharp cut-off in molecular sieving. On the other hand, this flexibility is helpful in matching the thermal expansion coefficients between membrane and support.
- The performance of MOF membranes can be predicted and designed by using the rough estimate “membrane selectivity = adsorption selectivity x diffusion selectivity” [8].
- Some MOFs are unstable without solvent, whereas for others – like ZIFs – a permanent porosity at temperatures 300 – 400°C, even in the presence of steam, is reported. When using MOFs as catalyst, oxidative degradation becomes an issue.
- Whereas mixed matrix membranes of zeolites/polymers showed no remarkable progress, MOF based mixed matrix membranes show revolutionary improvements since hydrophobic MOFs can interact perfectly with hydrophobic polymer matrices.

the synthesis tools

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Keywords: zeolite membrane, MOF membrane, molecular sieve membrane, gas separation