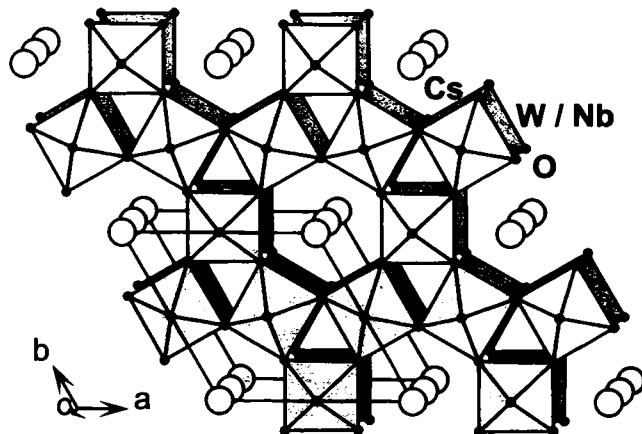


# Crystal structure of caesium niobium tungsten bronzes, $\text{Cs}_{0.23}(\text{Nb}_{0.09}\text{W}_{0.91})\text{O}_3$ and $\text{Cs}_{0.29}(\text{Nb}_{0.10}\text{W}_{0.90})\text{O}_3$

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## Abstract

$\text{Cs}_{0.23}\text{Nb}_{0.09}\text{O}_3\text{W}_{0.91}$  (1) hexagonal,  $P6_3/mcm$  (No. 193),  
 $a = 7.3998(2)$  Å,  $c = 7.5732(2)$  Å,  $V = 359.1$  Å<sup>3</sup>,  $Z = 6$ ,  
 $wR(P) = 0.062$ ,  $R(P) = 0.044$ ,  $R(I) = 0.023$ ,  $R(F) = 0.012$ ,  $T = 293$  K.

$\text{Cs}_{0.29}\text{Nb}_{0.10}\text{O}_3\text{W}_{0.90}$  (2), hexagonal,  $P6_3/mcm$  (No. 193),  
 $a = 7.3992(2)$  Å,  $c = 7.5867(2)$  Å,  $V = 359.7$  Å<sup>3</sup>,  $Z = 6$ ,  
 $wR(P) = 0.080$ ,  $R(P) = 0.057$ ,  $R(I) = 0.028$ ,  $R(F) = 0.014$ ,  $T = 293$  K.

## Source of material

The starting materials were  $\text{Cs}_2\text{WO}_4$ ,  $\text{WO}_3$ ,  $\text{WO}_2$  and  $\text{Nb}_2\text{O}_5$  (high quality reagent grade). Samples were prepared according to the following reaction:  $x/2 \text{Cs}_2\text{WO}_4 + (1-x-y/2) \text{WO}_3 + (x-y)/2 \text{WO}_2 + y/2 \text{Nb}_2\text{O}_5 \rightarrow \text{Cs}_x\text{Nb}_y\text{W}_{1-y}\text{O}_3$  ( $x = 0.25$  and  $0.30$ ;  $y = 0.1$ ). The reactants were mixed intimately in an agate mortar, taken into silica tubes, evacuated at room temperature for 2–3 hours, sealed, then heated in a Muffle furnace at a temperature of 1073 K for 7 days and quenched to room temperature. Both samples reveal single phase products of bluish black colour.

## Discussion

The crystal structures of  $\text{Cs}_x\text{Nb}_y\text{W}_{1-y}\text{O}_3$  ( $x = 0.23, 0.29$ ;  $y \approx 0.1$ ) are of hexagonal tungsten bronze type (HTB). Tungsten is substituted by niobium according to the predicted chemical reaction. A similar result was also obtained for the analogous  $\text{Rb}_x\text{Nb}_y\text{W}_{1-y}\text{O}_3$  ( $x = 0.3$  with  $y = 0.13$  and  $0.19$ ) [1]. The properties of the present compounds  $\text{Cs}_x\text{Nb}_y\text{W}_{1-y}\text{O}_3$  are also observed in close analogy to that reported for  $\text{Rb}_x\text{Nb}_y\text{W}_{1-y}\text{O}_3$ , revealing decreasing carrier concentration with increasing  $y$  and a possible metal insulator transition [2].

Rewriting the formula of tungsten bronzes as  $\text{M}_x(\text{W}_x^{5+}\text{W}_{1-x}^{6+})\text{O}_3$  it implies a substitution  $\text{Cs}_x(\text{W}_{x-y}^{5+}\text{Nb}_y^{5+}\text{W}_{1-x-y}^{6+})\text{O}_3$ . According to this, the number of charge carriers may be counted by the number of  $\text{W}^{5+}$  states, decreasing with decreasing  $x$  and increasing  $y$ .

## 1. Caesium niobium tungsten bronze, $\text{Cs}_{0.23}(\text{Nb}_{0.09}\text{W}_{0.91})\text{O}_3$

Table 1. Data collection and handling.

Powder:	bluish black
Wavelength:	$\text{Cu K}_\alpha$ radiation (1.54059 Å)
$\mu$ :	10.98 cm <sup>-1</sup>
Diffractometer:	Stoe STADI P, transmission, Debye-Scherrer mode
$2\theta_{\max}$ , stepwidth:	115.14°, 0.02°
$N(\text{points})_{\text{measured}}$ :	5258
$N(\text{hkl})_{\text{measured}}$ :	108
$N(\text{param})_{\text{refined}}$ :	29
Program:	Rietan-97 [3]

Table 2. Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

Atom	Site	Occ.	x	y	z	B <sub>iso</sub>
Cs	2b	0.679(5)	0	0	0	1.7(1)
Nb	6g	0.09(1)	0.4766(1)	0	1/4	0.48(3)
W	6g	0.91	0.4766	0	1/4	0.48
O(1)	6f		1/2	0	0	1.7(2)
O(2)	12j		0.418(1)	0.211(1)	1/4	1.7(2)

## 2. Caesium niobium tungsten bronze, $\text{Cs}_{0.29}(\text{Nb}_{0.10}\text{W}_{0.90})\text{O}_3$

Table 3. Data collection and handling.

Powder:	bluish black
Wavelength:	$\text{Cu K}_\alpha$ radiation (1.54059 Å)
$\mu$ :	11.61 cm <sup>-1</sup>
Diffractometer:	Stoe STADI P, transmission, Debye-Scherrer mode
$2\theta_{\max}$ , stepwidth:	119.98°, 0.02°
$N(\text{points})_{\text{measured}}$ :	5500
$N(\text{hkl})_{\text{measured}}$ :	116
$N(\text{param})_{\text{refined}}$ :	29
Program:	Rietan-97 [3]

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**Table 4.** Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

Atom	Site	Occ.	x	y	z	B <sub>iso</sub>
Cs(1)	2b	0.864(5)	0	0	0	1.5(1)
Nb(1)	6g	0.10(1)	0.4816(1)	0	1/4	0.36(3)
W(1)	6g	0.90	0.4816	0	1/4	0.36
O(1)	6f		1/2	0	0	0.9(2)
O(2)	12j		0.414(1)	0.205(2)	1/4	0.9(2)

**References**

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